Errors in **Introduction to Quantum Mechanics in Chemistry** by Mark Ratner and George Schatz

Preface

Page x, line 8, "Problems 2.12" should be "Problems 2.11"

Chapter 1

Exercise 1.5, part 2, add to the parenthetic expression the sentence: Assume that the intensity refers to photons per unit area.

Chapter 2

Page 27, change m<sub>i</sub> to m<sub>i</sub> before Eq. 2.9b.

Page 35, two lines before Eq. (2.41), replace "any Hermitian operator" by "any operator corresponding to a physical observable".

Chapter 3

Eq. 3.21 should be:

$$< x^2 > = \frac{a^2}{3} \left( 1 - \frac{3}{2\pi^2 n^2} \right)$$

Eq. 3.22 should be:

$$<\Delta x> = \sqrt{\langle x^2> - \langle x>^2} = \sqrt{\langle (x-\langle x>)^2} = \sqrt{\frac{a^2}{12}(1-\frac{6}{\pi^2n^2})}$$

Eq. 3.26 should be:

$$(\Delta x)(\Delta p) \cong \frac{hn}{2a} \sqrt{\frac{a^2}{12} (1 - \frac{6}{\pi^2 n^2})} = \frac{nh}{4\sqrt{3}} \sqrt{1 - \frac{6}{\pi^2 n^2}} > \frac{\hbar}{2}$$

Page 48, second line, replace "(FEMO) imodel s a simple model" with "(FEMO) model is a simple model"

problem 3.10 It would be best to add a hint: The area of the molecule is 5.09Å<sup>2</sup>.

Chapter 4

Eq. 4.58 func  $\{\partial \text{ over } \{\partial u\}\}\$  should be func  $\{\partial \text{ over } \{\partial \text{ theta}\}\}\$ 

Chapter 7

Eq. 7.19

## 2.8377 should be 2.8477

Chapter 10

problem 10.4 It is better to take K=4.375 than 1.75

Chapter 12

problem 12.9 (d) "effective" should be "net"

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